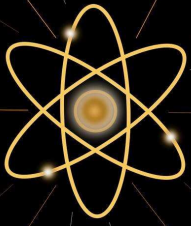


- 1) $2CO + O_2 \rightarrow 2CO_2$
- 2) $CO_2 + CaO \rightarrow CaCO_3$
- 3) $CuO + CO \rightarrow Cu + CO_2$
- 4) $Fe_2O_3 + 3CO \rightarrow 2Fe + 3CO_2$
- 5) $2HCl + CaO \rightarrow CaCl_2 + H_2O$
- 6) $HCl + NH_3 \rightarrow NH_4Cl$
- 7) $2HCl + MgO \rightarrow MgCl_2 + H_2O$
- 8) $2H_2S + 3O_2 \rightarrow 2SO_2 + 2H_2O$
- 9) $2SO_2 + O_2 \rightarrow 2SO_3$
- 10) $2NO + O_2 \rightarrow 2NO_2$

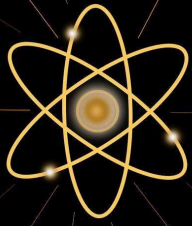
$$\Delta H_{reak.} = \sum \Delta H_{prod.} - \sum \Delta H_{reak.}$$

Entalpija



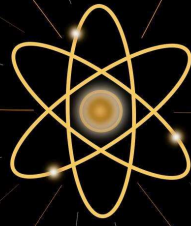
5kj/mol

Entalpija



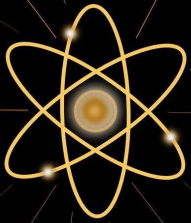
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Entalpija



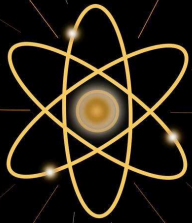
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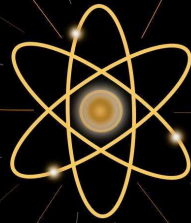
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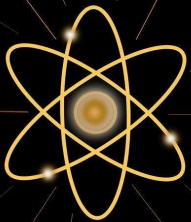
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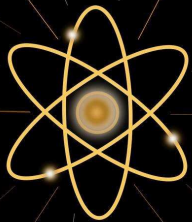
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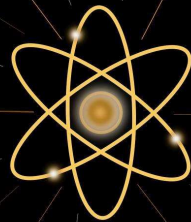
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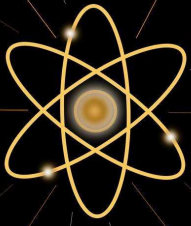
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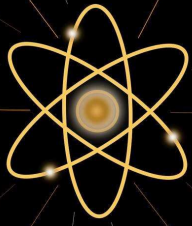
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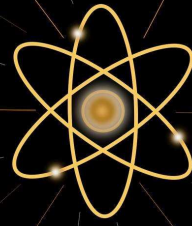
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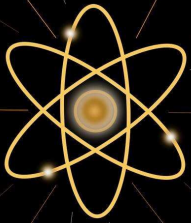
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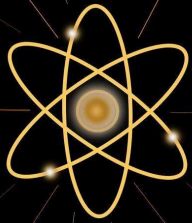
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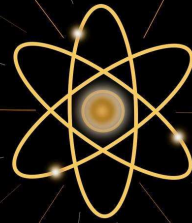
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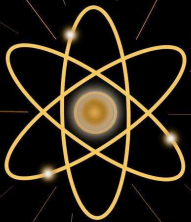
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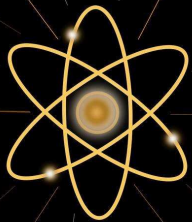
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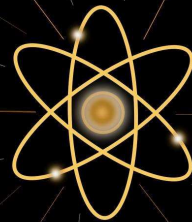
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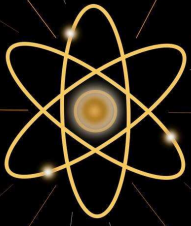
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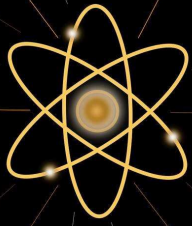
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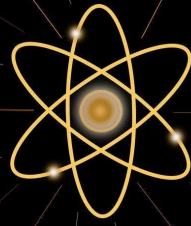
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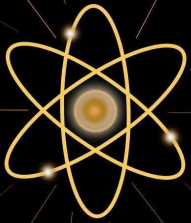
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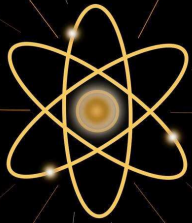
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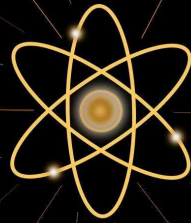
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Enthalpija



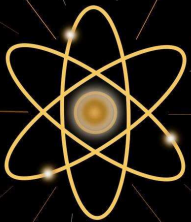
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Enthalpija



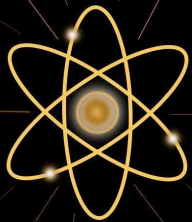
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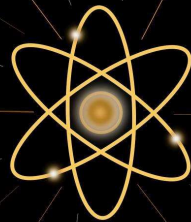
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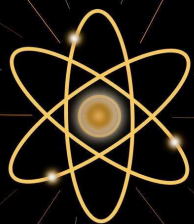
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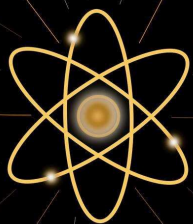
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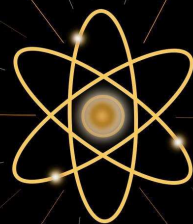
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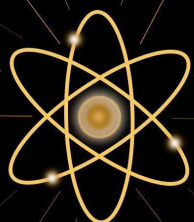
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Enthalpija



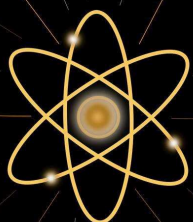
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Enthalpija



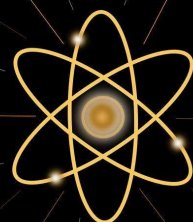
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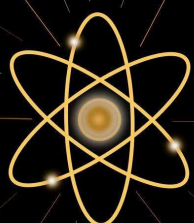
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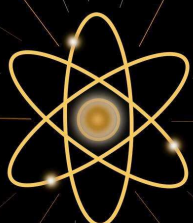
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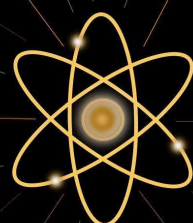
10kj/mol

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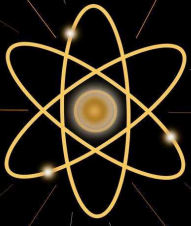
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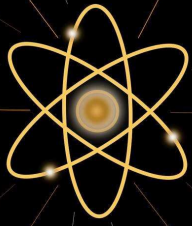
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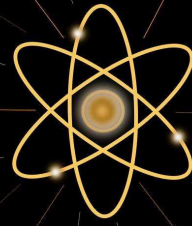
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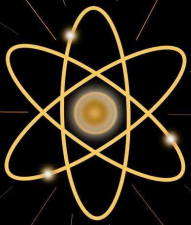
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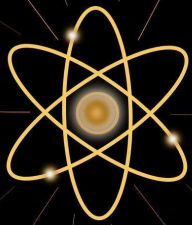
10kj/mol

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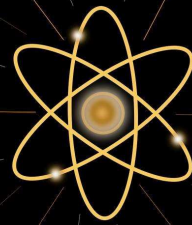
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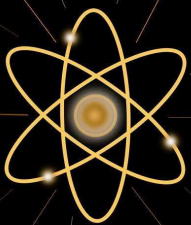
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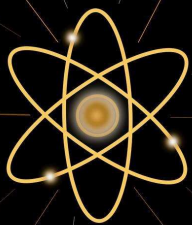
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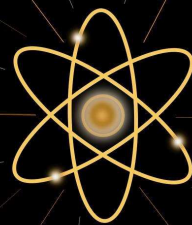
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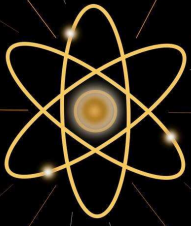
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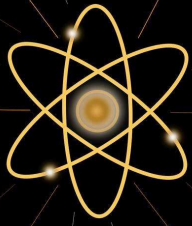
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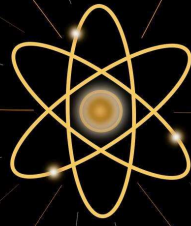
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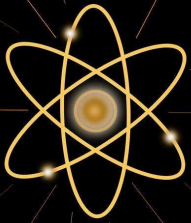
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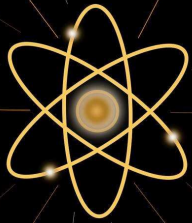
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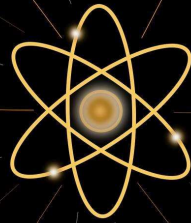
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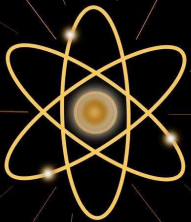
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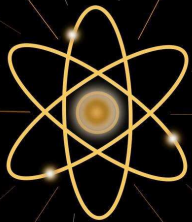
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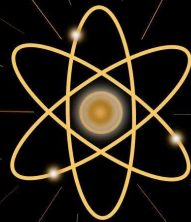
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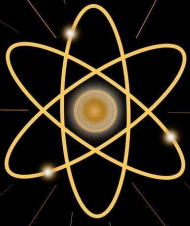
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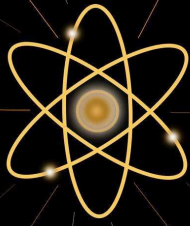
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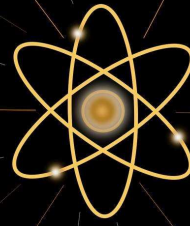
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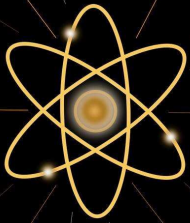
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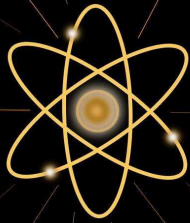
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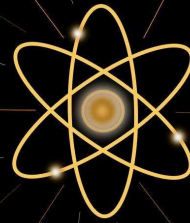
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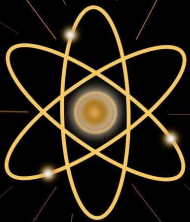
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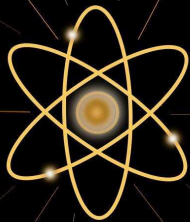
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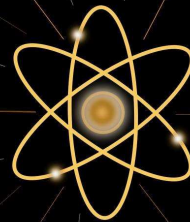
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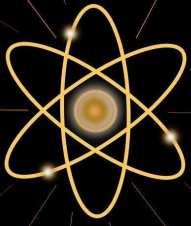
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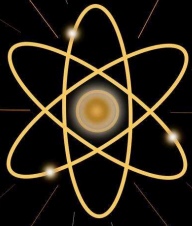
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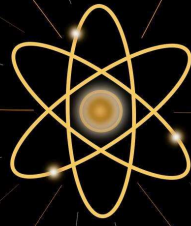
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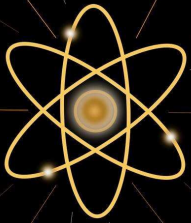
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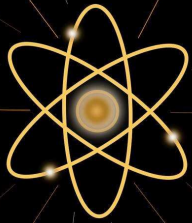
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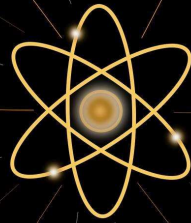
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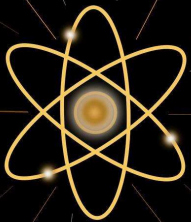
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Enthalpija



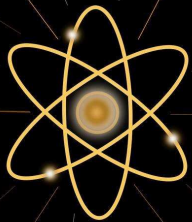
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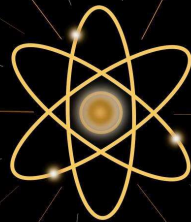
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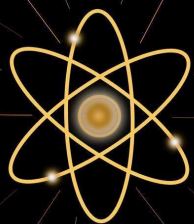
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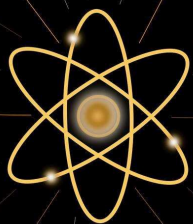
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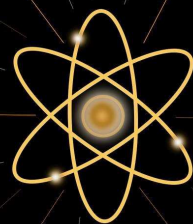
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Entalpija



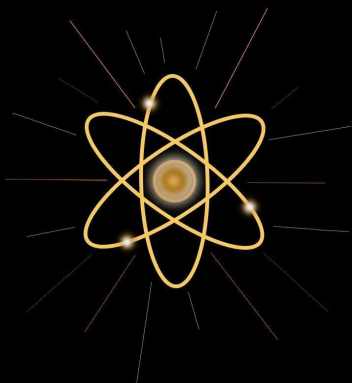
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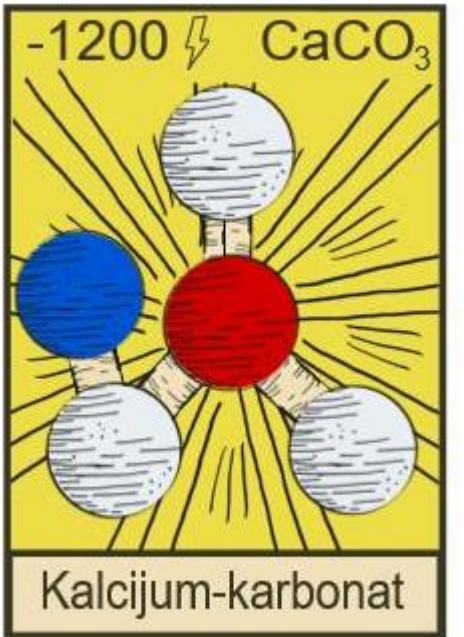
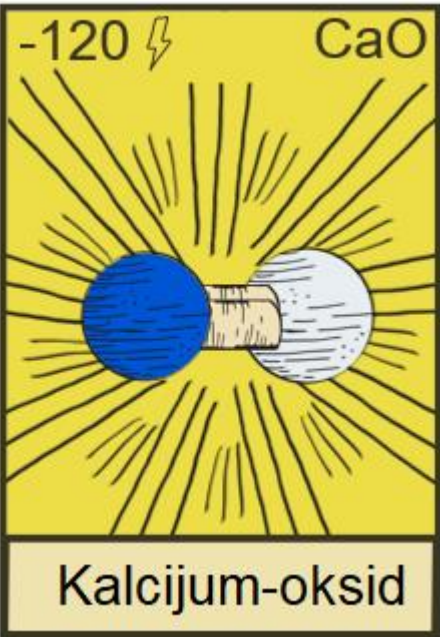
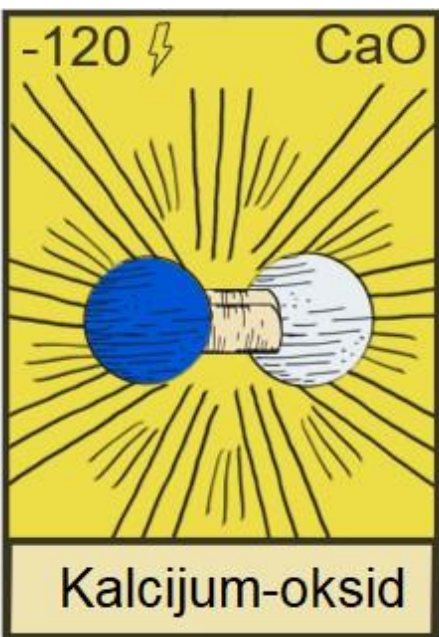
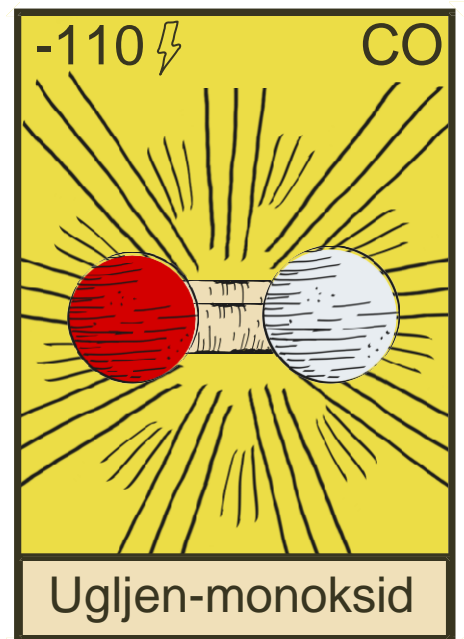
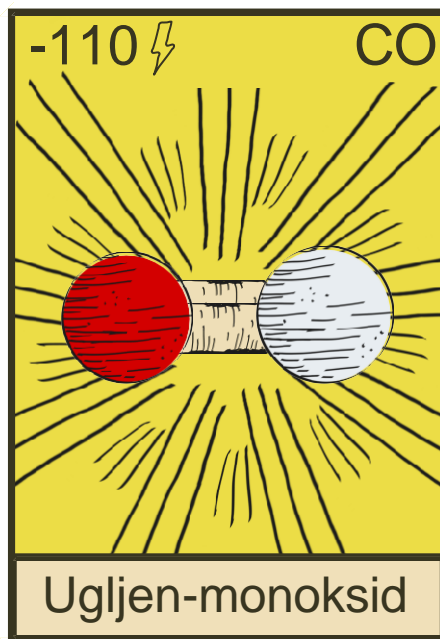
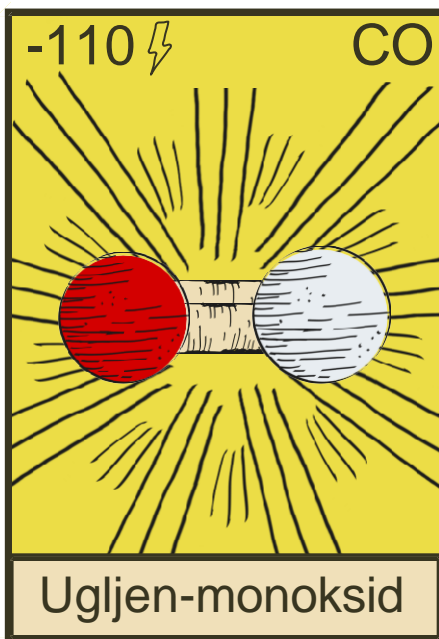
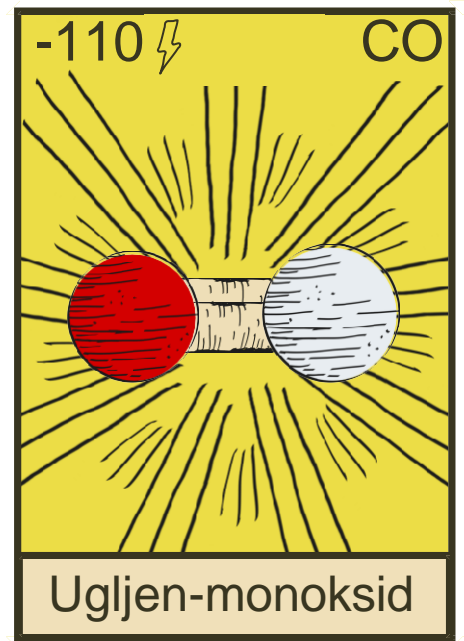
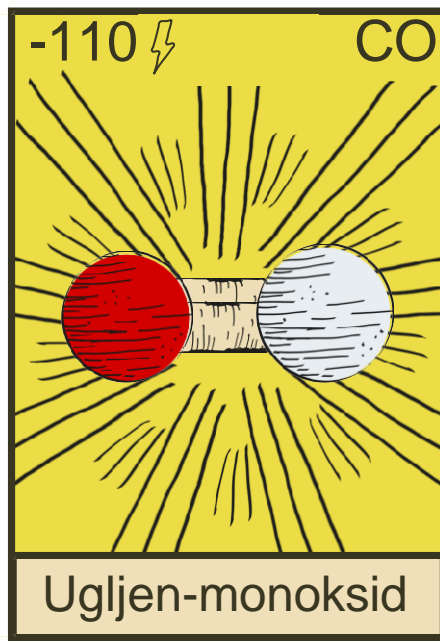
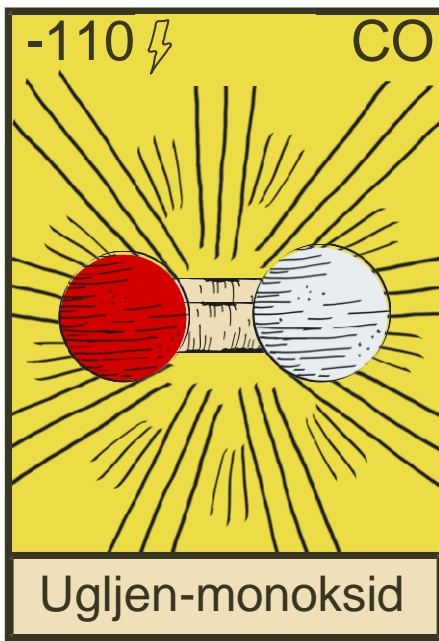


100kj/mol

Entalpija

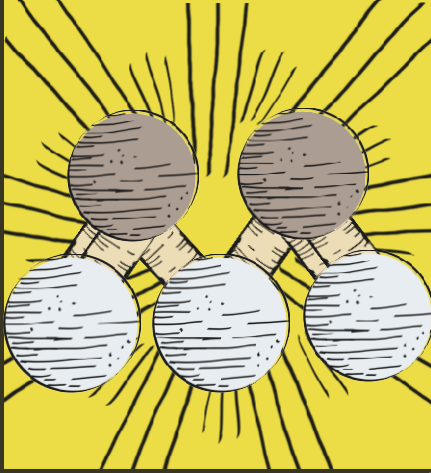


100kj/mol



-820 ⚡

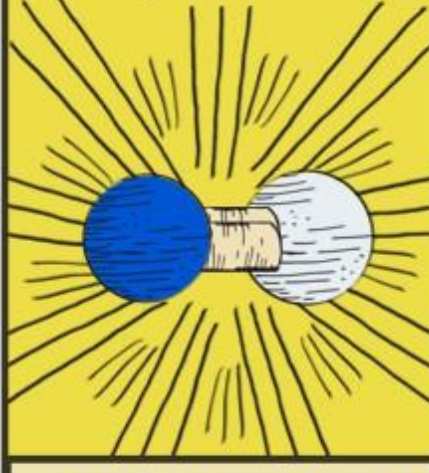
Fe_2O_3



Gvožđe(III)-oksid

-120 ⚡

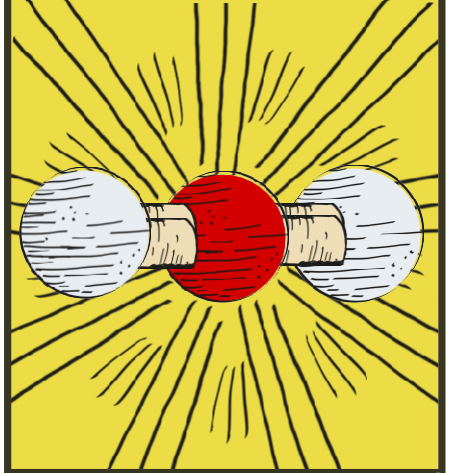
CaO



Kalcijum-oksid

-395 ⚡

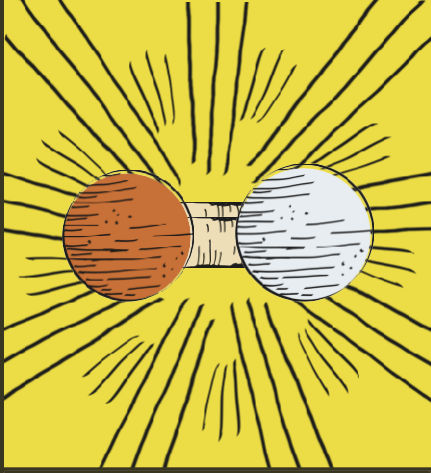
CO_2



Ugljen-dioksid

-155 ⚡

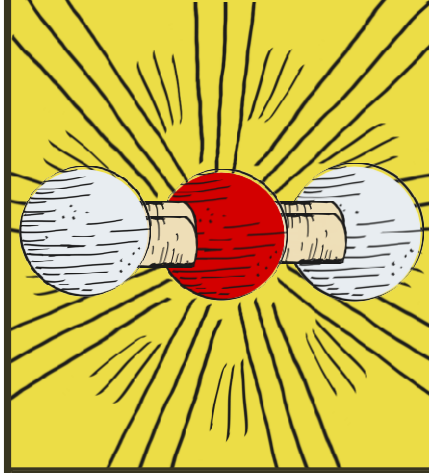
CuO



Bakar(II)-oksid

-395 ⚡

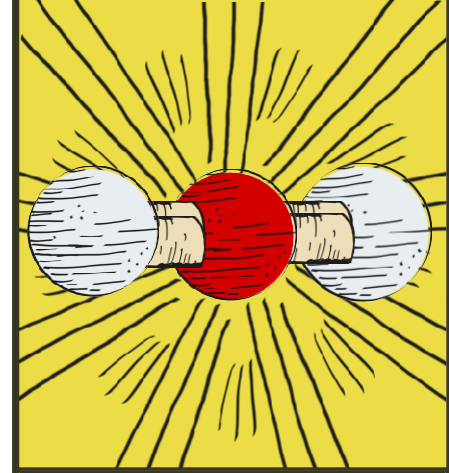
CO_2



Ugljen-dioksid

-395 ⚡

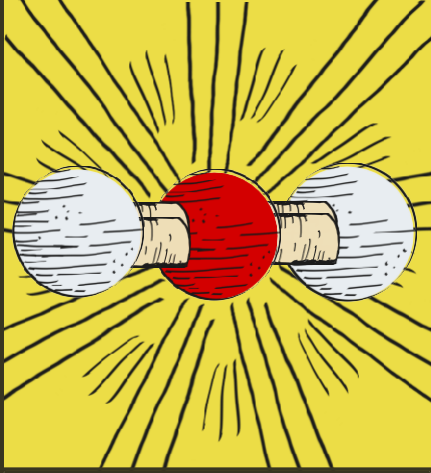
CO_2



Ugljen-dioksid

-395 ⚡

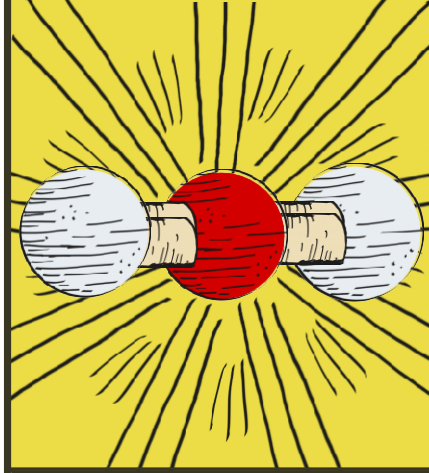
CO_2



Ugljen-dioksid

-395 ⚡

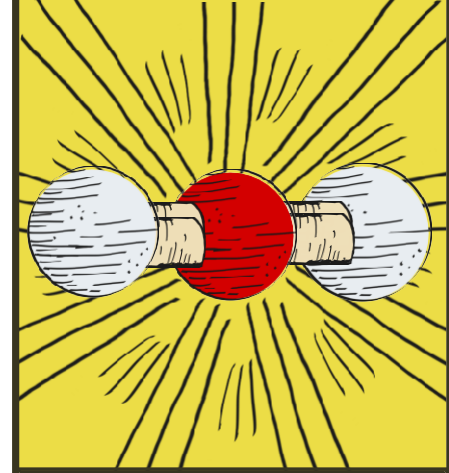
CO_2



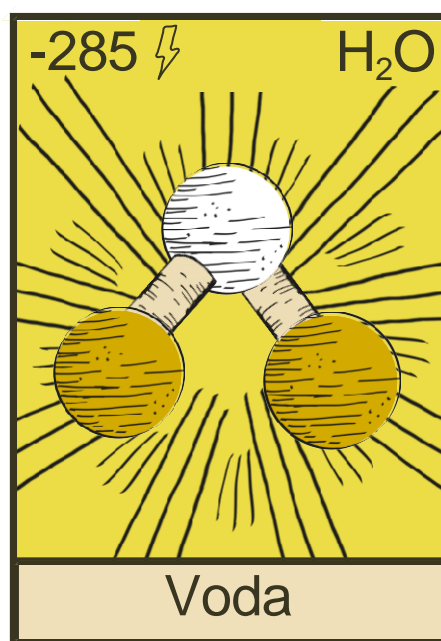
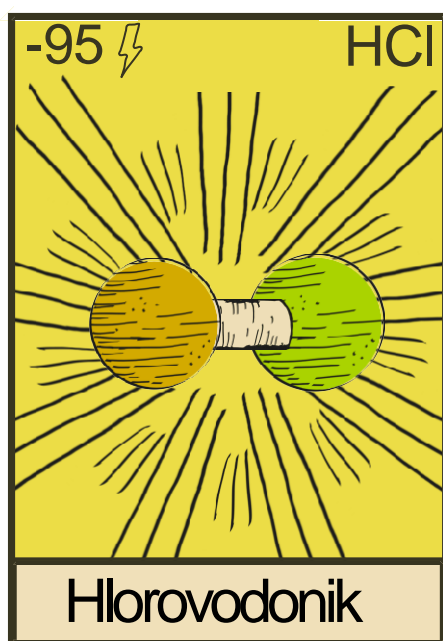
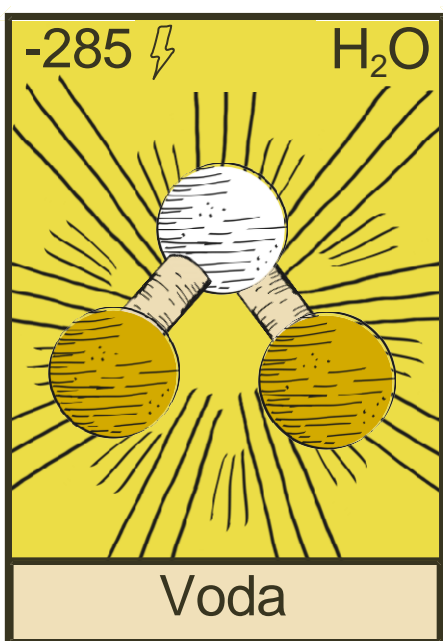
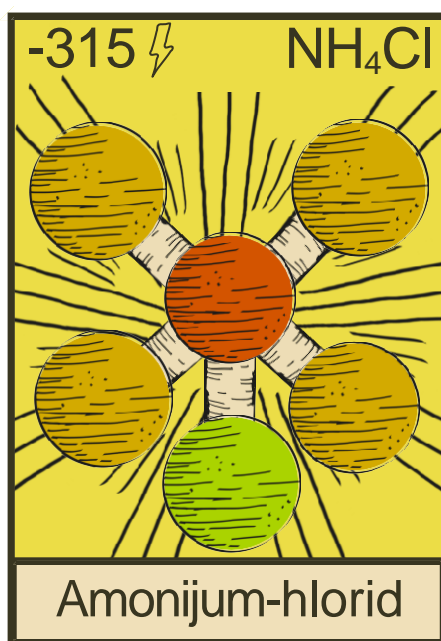
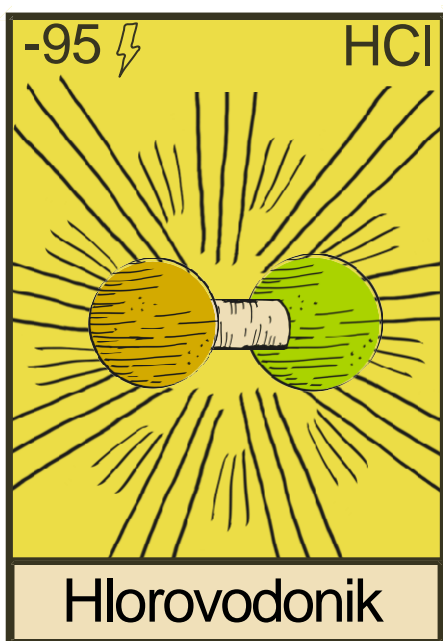
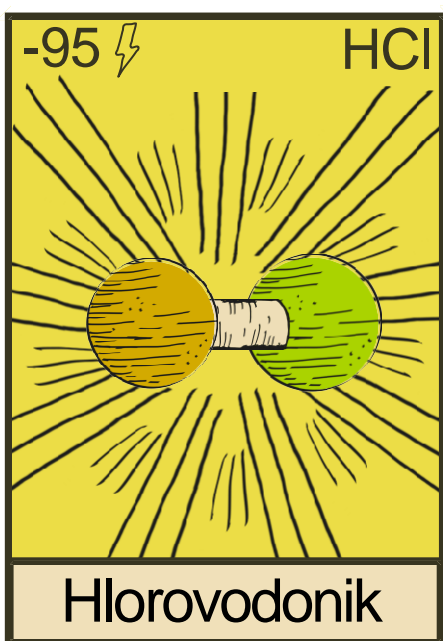
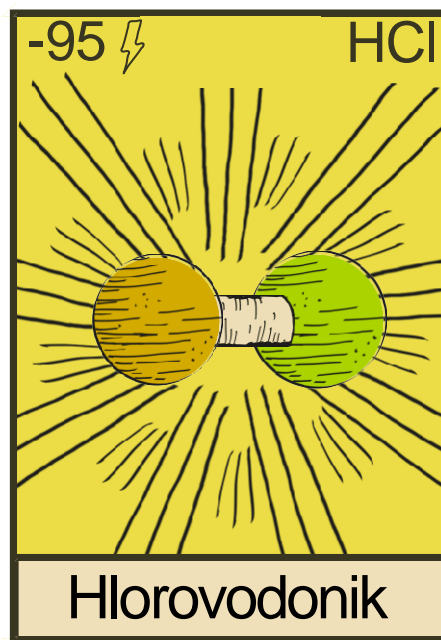
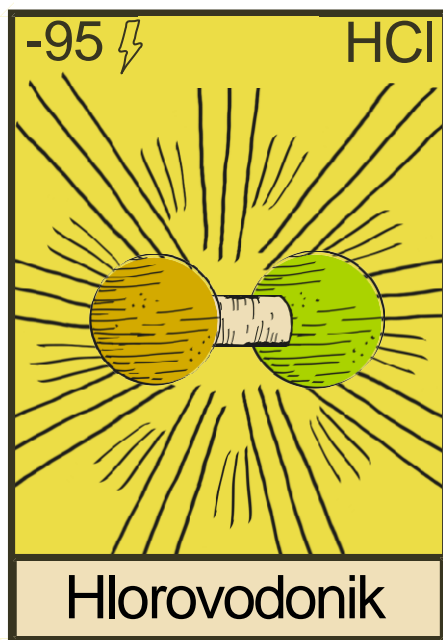
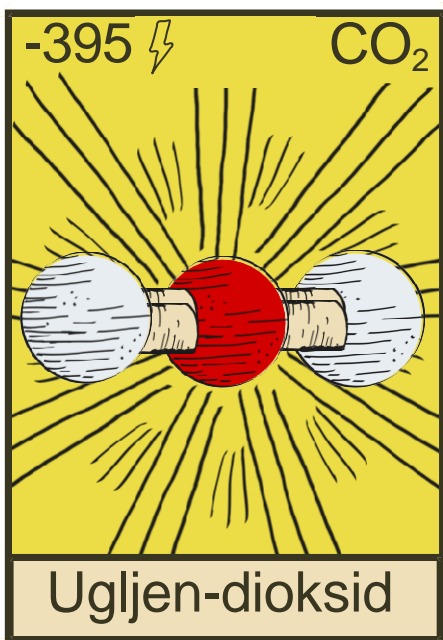
Ugljen-dioksid

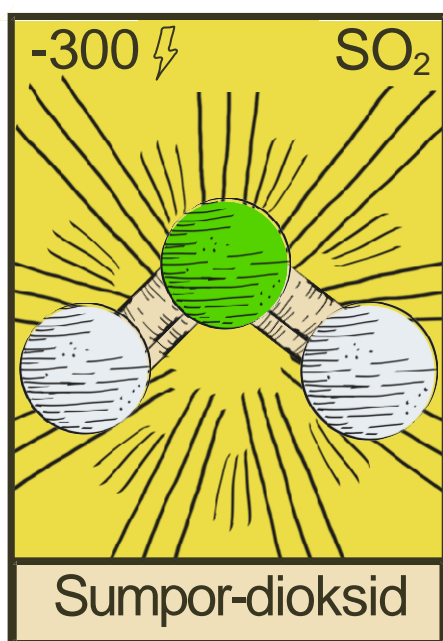
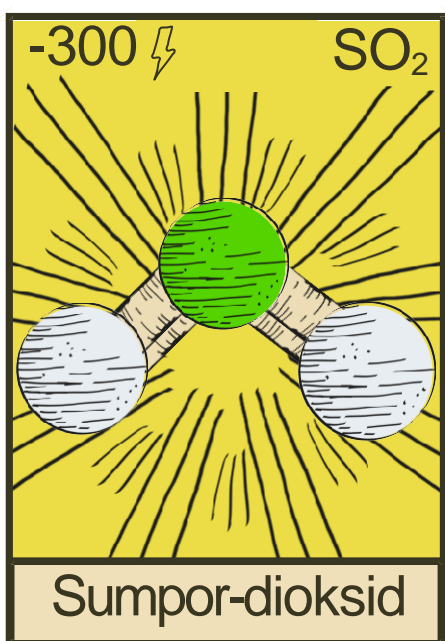
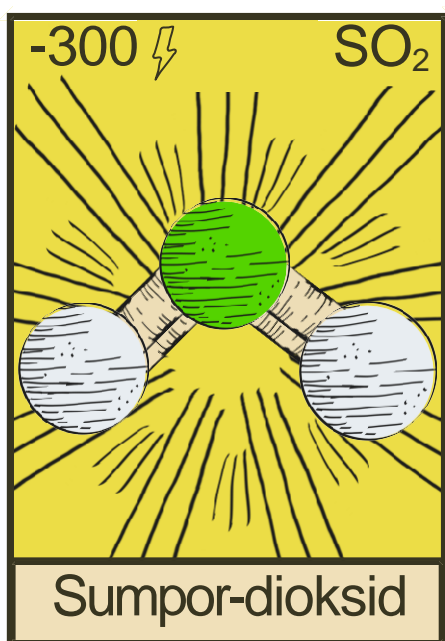
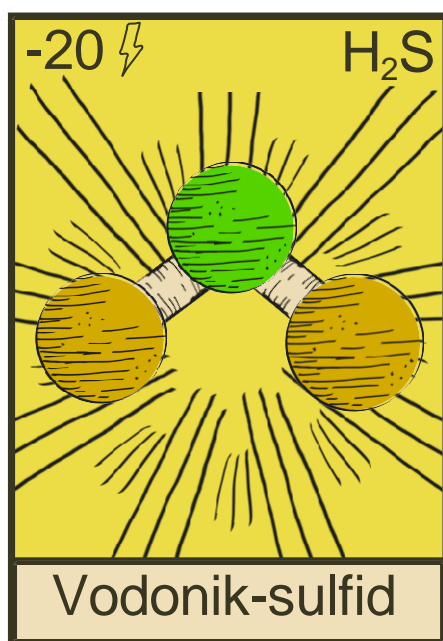
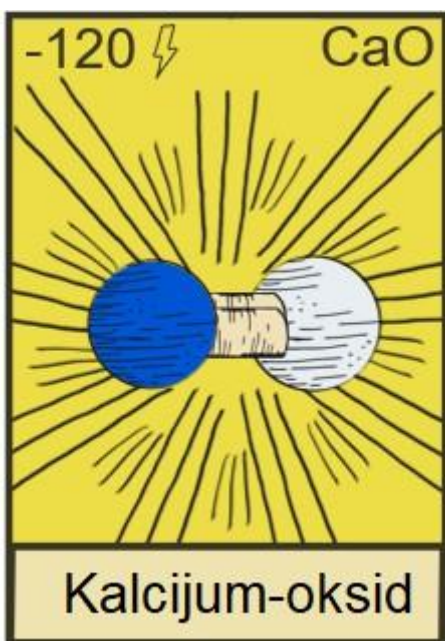
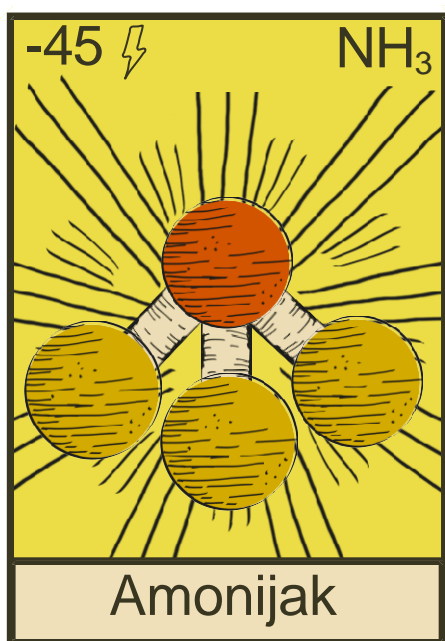
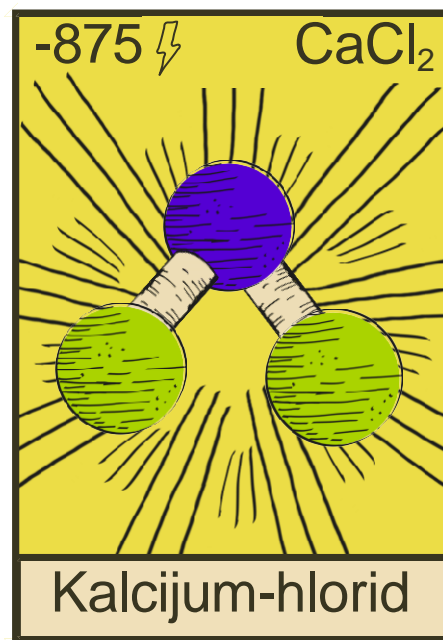
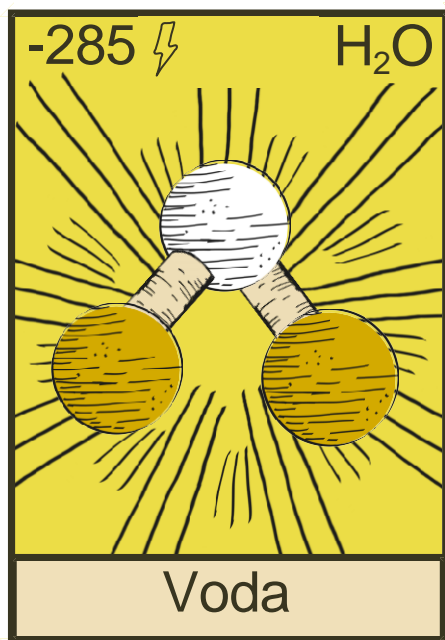
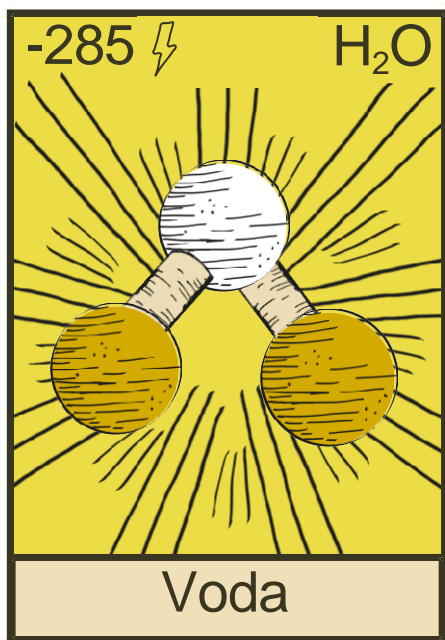
-395 ⚡

CO_2



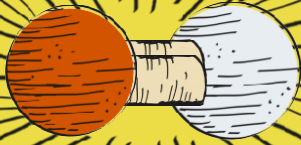
Ugljen-dioksid





+90 ⚡

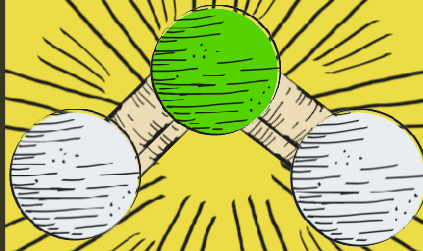
NO



Azot-monoksid

-300 ⚡

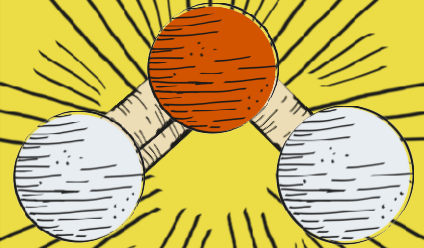
SO₂



Sumpor-dioksid

+35 ⚡

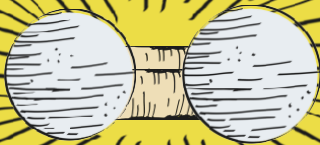
NO₂



Azot-dioksid

0 ⚡

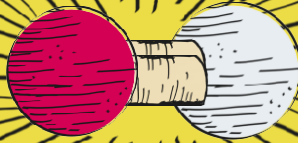
O₂



Kiseonik

-600 ⚡

MgO



Magnezijum-oksidi

0 ⚡

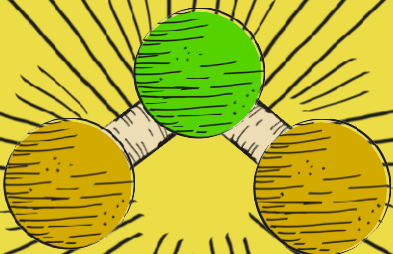
Fe



Gvožđe

-20 ⚡

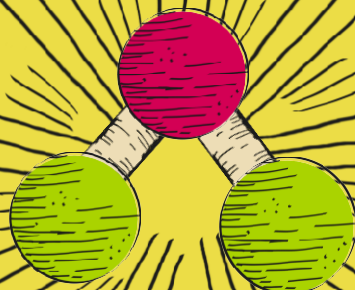
H₂S



Vodonik-sulfid

-625 ⚡

MgCl₂



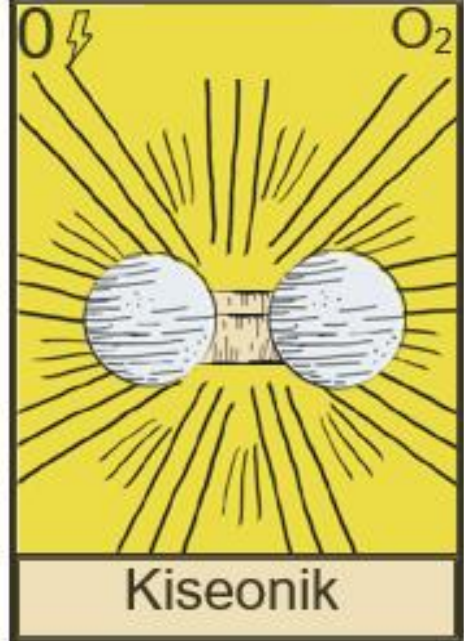
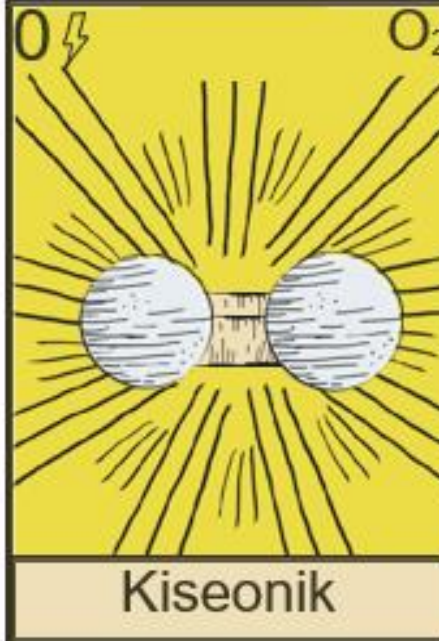
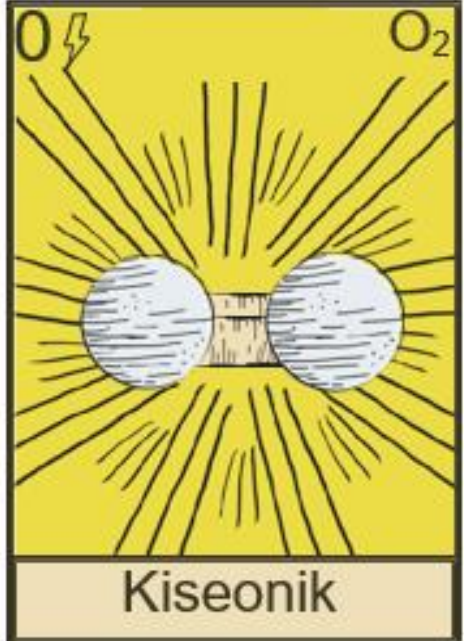
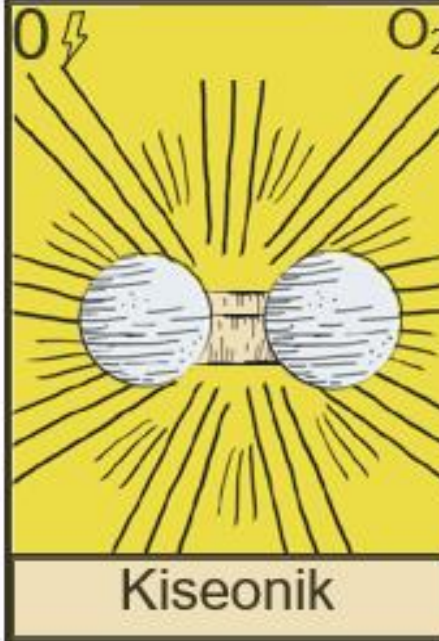
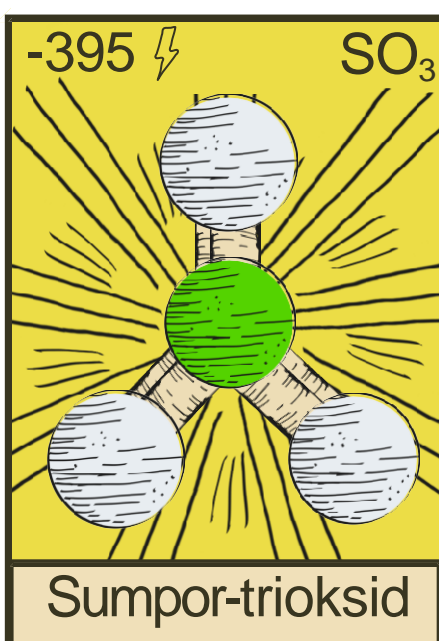
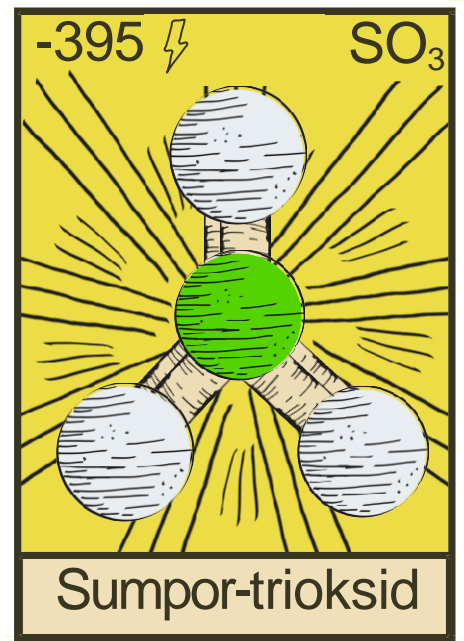
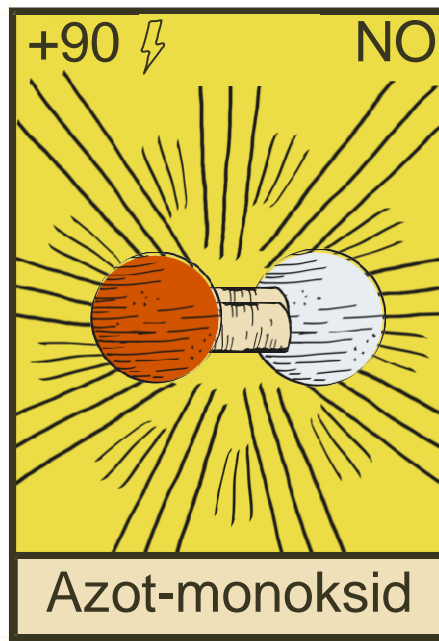
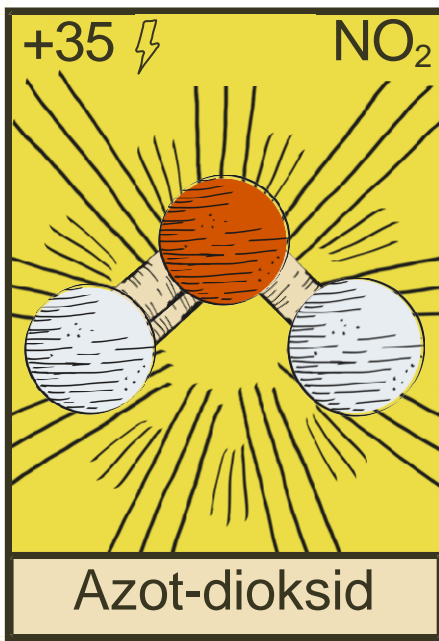
Magnezijum-hlorid

0 ⚡

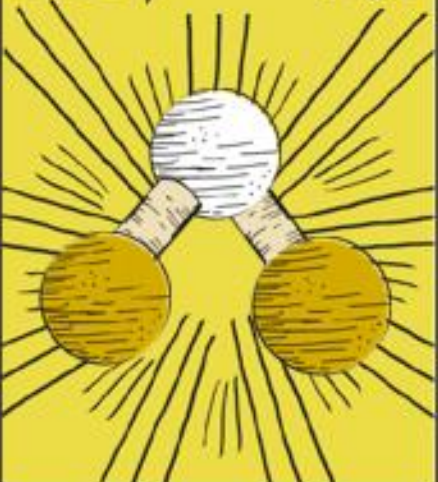
Cu



Bakar




-285 ⚡ H₂O



Voda

The diagram shows a ball-and-stick model of a water molecule (H₂O). It consists of one white sphere (oxygen) at the top, connected by two grey rods to two yellow spheres (hydrogen) at the bottom. The entire molecule is set against a yellow background with black radiating lines. A lightning bolt symbol is positioned to the left of the number -285.

0 ⚡ Fe



Gvožđe

The diagram shows a ball-and-stick model of an iron atom (Fe). It consists of a single grey sphere (iron) in the center. The entire atom is set against a yellow background with black radiating lines. A lightning bolt symbol is positioned to the left of the number 0.

