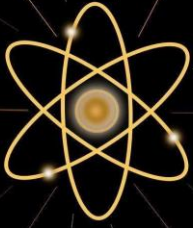


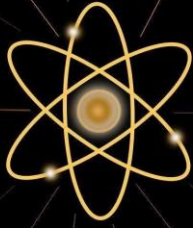
$$\Delta H_{reak.} = \sum \Delta H_{prod.} - \sum \Delta H_{reak.}$$

Enthalpija



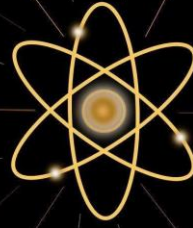
5kj/mol

Enthalpija



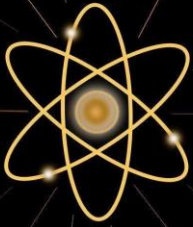
5kj/mol

Enthalpija



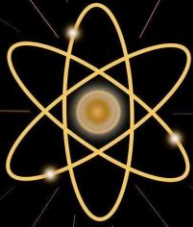
5kj/mol

Enthalpija



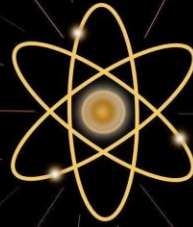
5kj/mol

Enthalpija



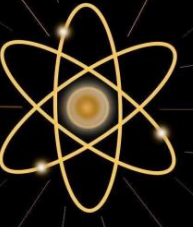
5kj/mol

Enthalpija



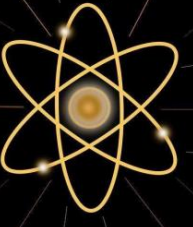
5kj/mol

Enthalpija



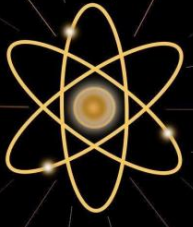
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Enthalpija



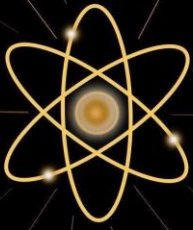
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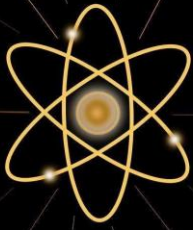
5kj/mol

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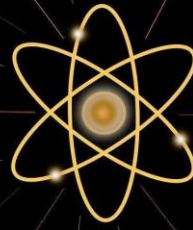
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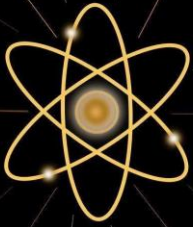
5kj/mol

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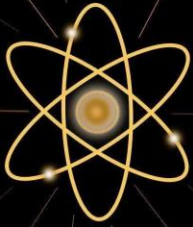
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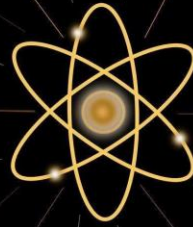
5kj/mol

Enthalpija



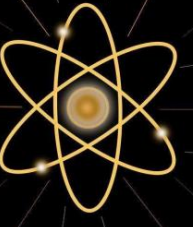
20kj/mol

Enthalpija



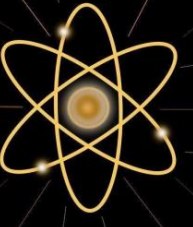
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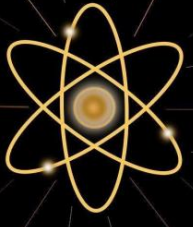
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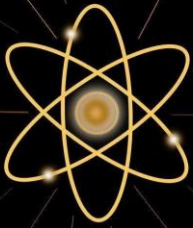
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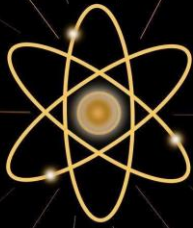
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Enthalpija



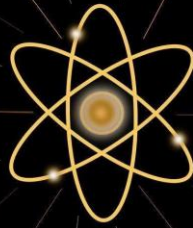
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Enthalpija



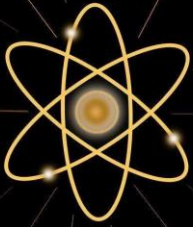
20kj/mol

Enthalpija



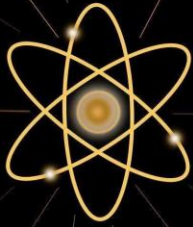
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Enthalpija



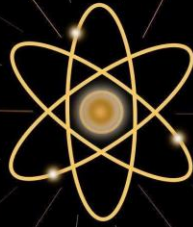
20kj/mol

Enthalpija



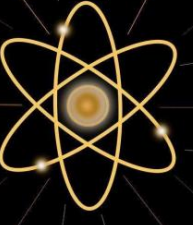
20kj/mol

Enthalpija



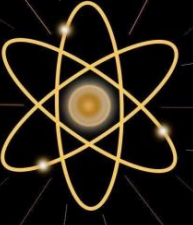
20kj/mol

Enthalpija



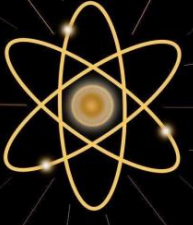
20kj/mol

Enthalpija



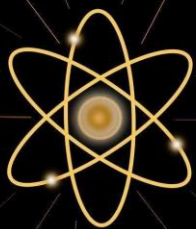
20kj/mol

Enthalpija



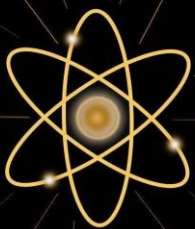
20kj/mol

Enthalpija



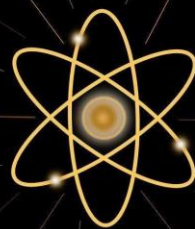
20kj/mol

Enthalpija



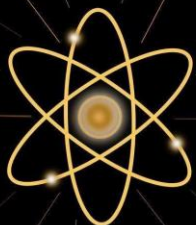
10kj/mol

Enthalpija



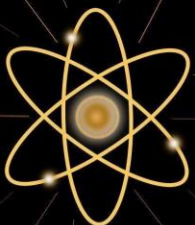
10kj/mol

Enthalpija



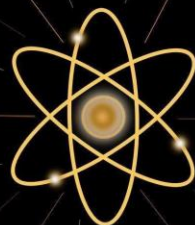
10kj/mol

Enthalpija



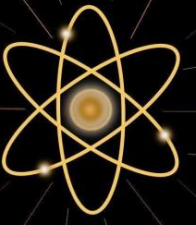
10kj/mol

Enthalpija



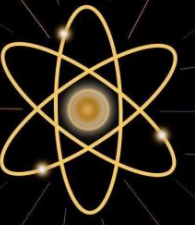
10kj/mol

Enthalpija



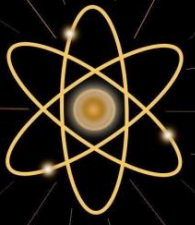
10kj/mol

Enthalpija



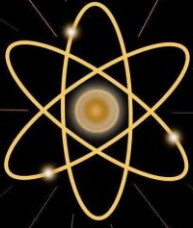
10kj/mol

Enthalpija



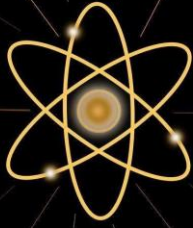
10kj/mol

Enthalpija



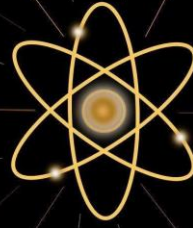
10kj/mol

Enthalpija



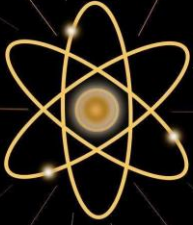
10kj/mol

Enthalpija



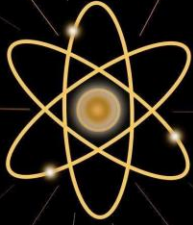
10kj/mol

Enthalpija



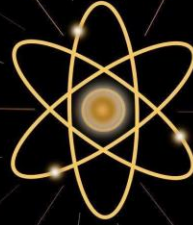
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Enthalpija



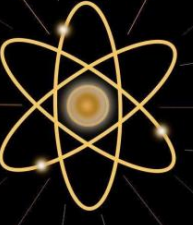
50kj/mol

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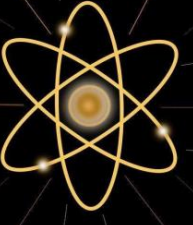
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Enthalpija



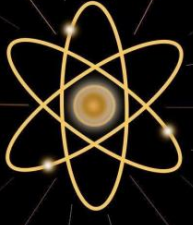
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Enthalpija



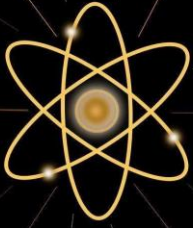
50kj/mol

Enthalpija



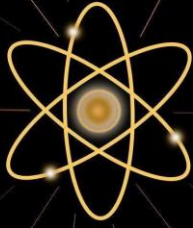
50kj/mol

Enthalpija



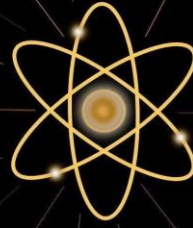
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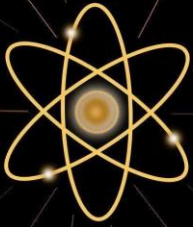
50kj/mol

Enthalpija



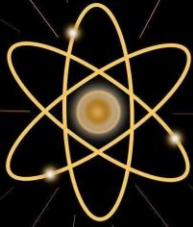
50kj/mol

Enthalpija



50kj/mol

Enthalpija



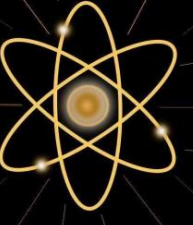
50kj/mol

Enthalpija



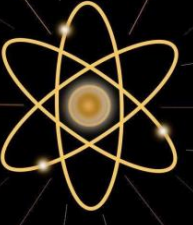
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Enthalpija



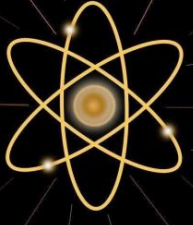
50kj/mol

Enthalpija



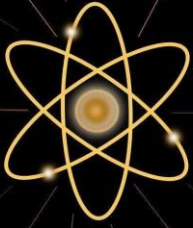
50kj/mol

Enthalpija



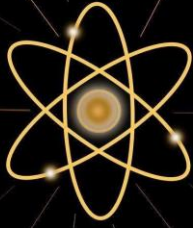
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Enthalpija



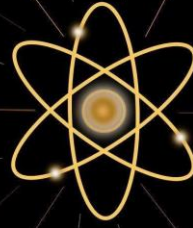
50kj/mol

Enthalpija



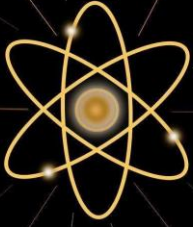
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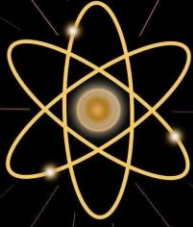
50kj/mol

Enthalpija



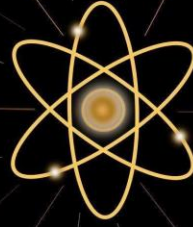
50kj/mol

Enthalpija



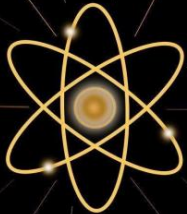
50kj/mol

Enthalpija



50kj/mol

Enthalpija



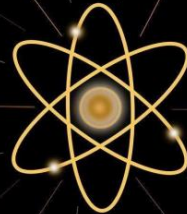
100kj/mol

Enthalpija



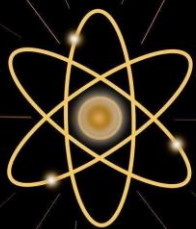
100kj/mol

Enthalpija



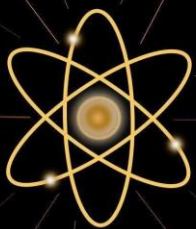
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Enthalpija



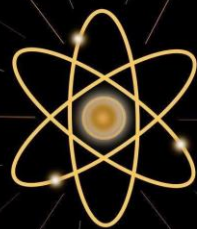
100kj/mol

Enthalpija



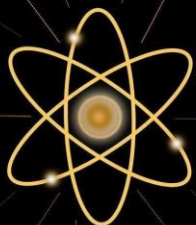
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Enthalpija



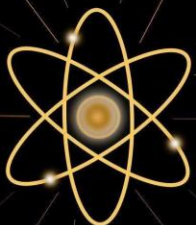
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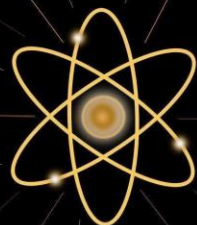
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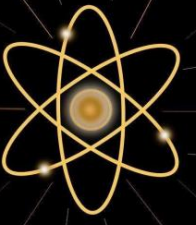
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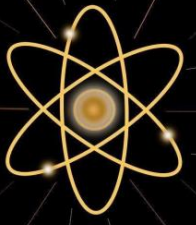
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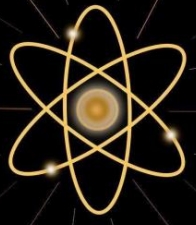
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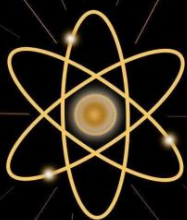
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Enthalpija



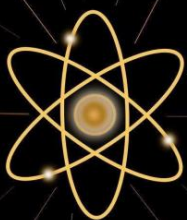
100kj/mol

Enthalpija



100kj/mol

Enthalpija



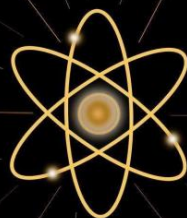
100kj/mol

Enthalpija

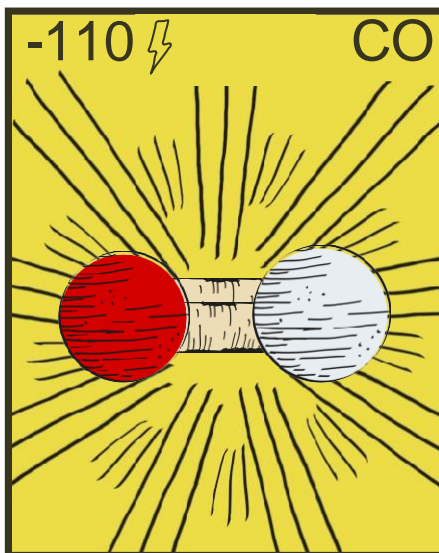


100kj/mol

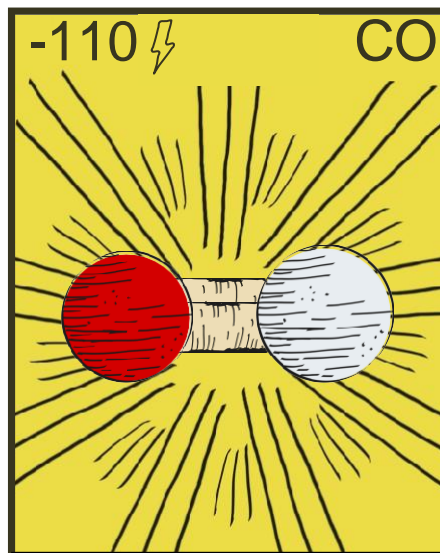
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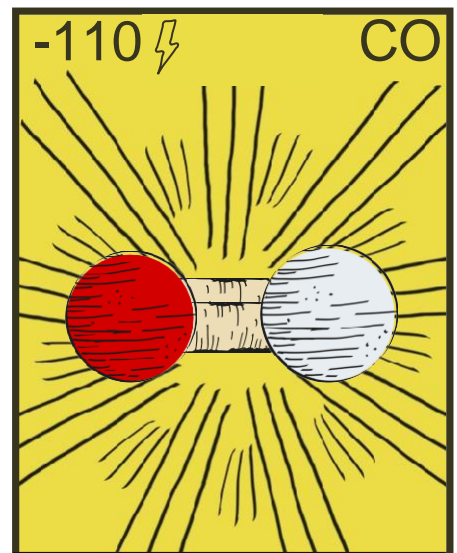
100kj/mol



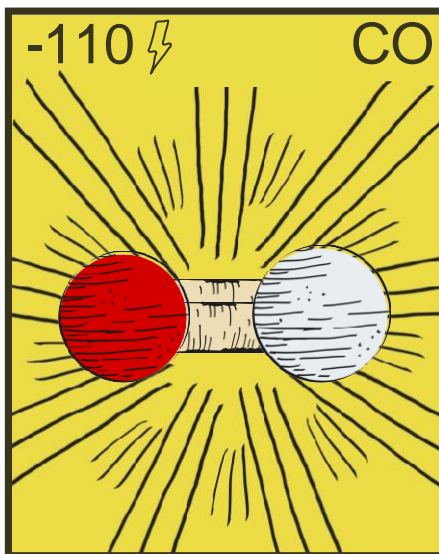
Carbon (II) oxide



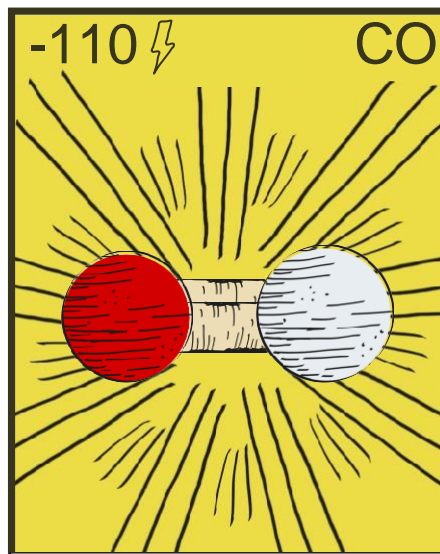
Carbon (II) oxide



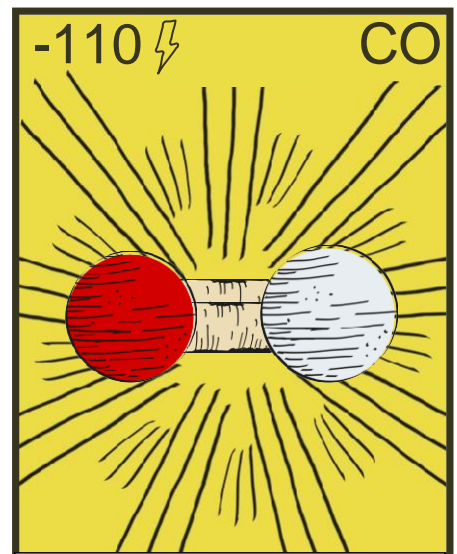
Carbon (II) oxide



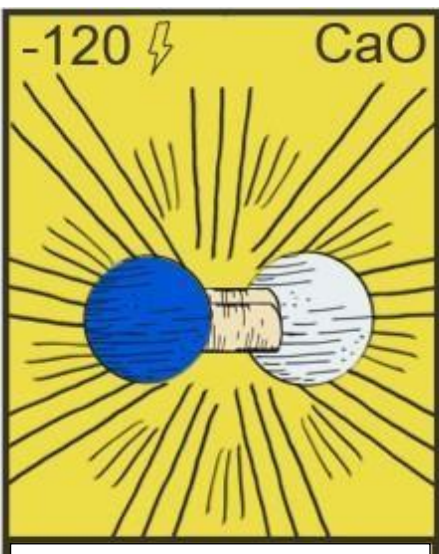
Carbon (II) oxide



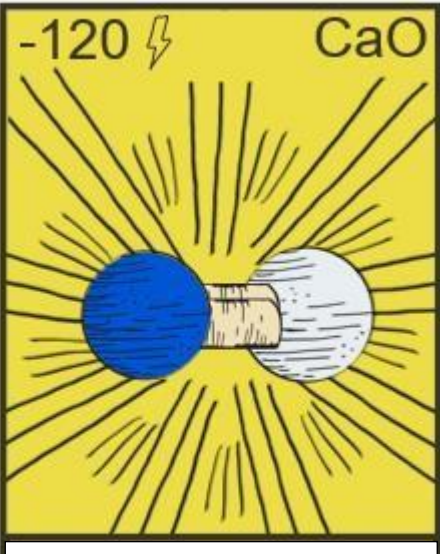
Carbon (II) oxide



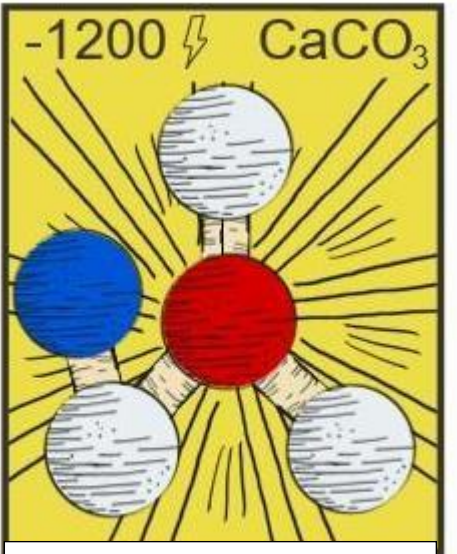
Carbon (II) oxide



Calcium oxide

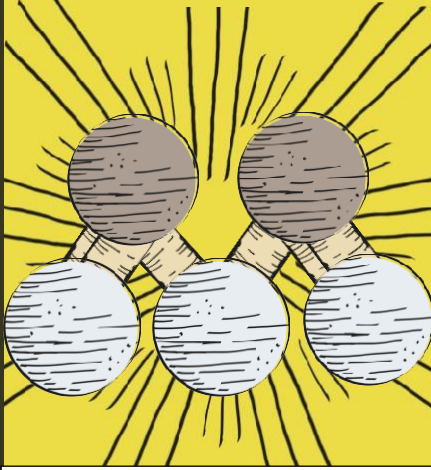


Calcium oxide



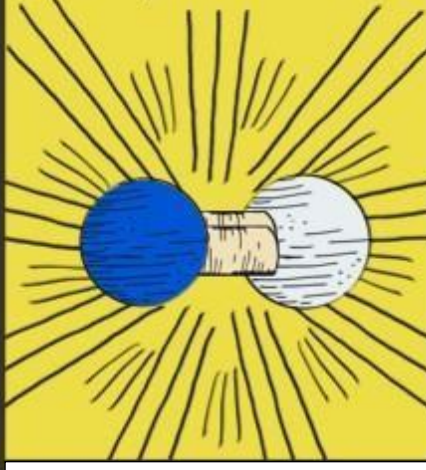
Calcium carbonate

-820 ⚡ Fe₂O₃



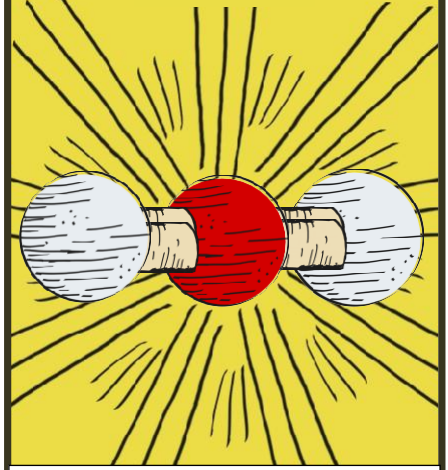
Iron (III) oxide

-120 ⚡ CaO



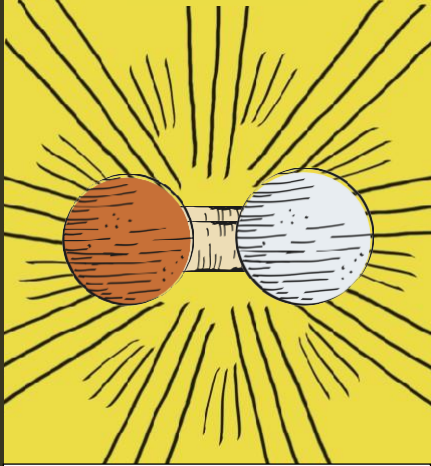
Calcium oxide

-395 ⚡ CO₂



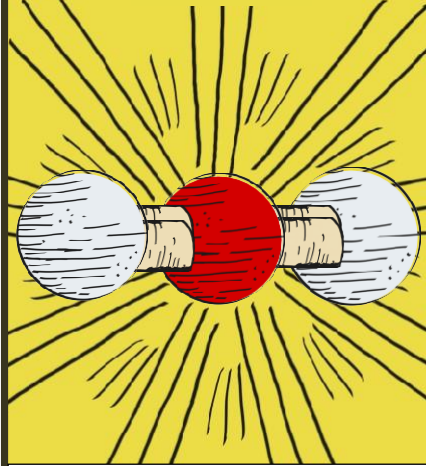
Carbon (IV) oxide

-155 ⚡ CuO



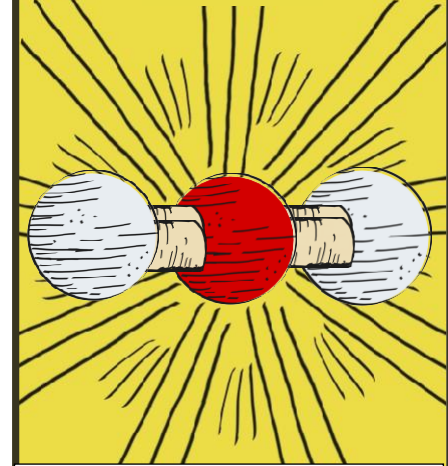
Copper (II) oxide

-395 ⚡ CO₂



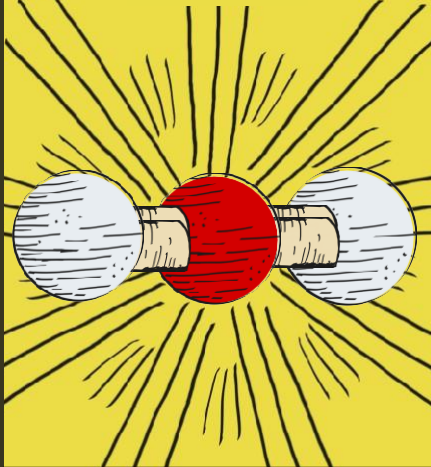
Carbon (IV) oxide

-395 ⚡ CO₂



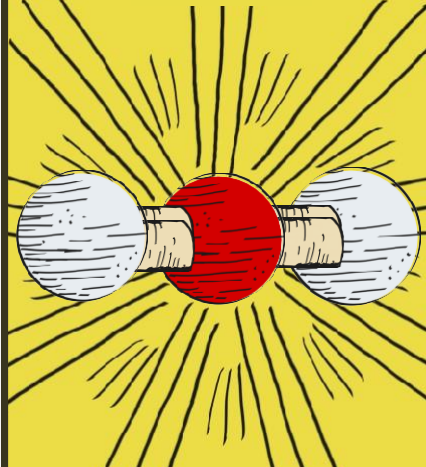
Carbon (IV) oxide

-395 ⚡ CO₂



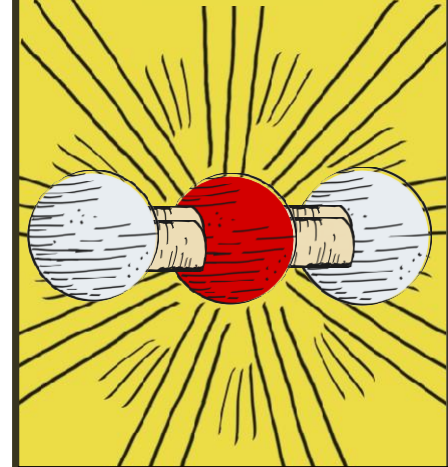
Carbon (IV) oxide

-395 ⚡ CO₂



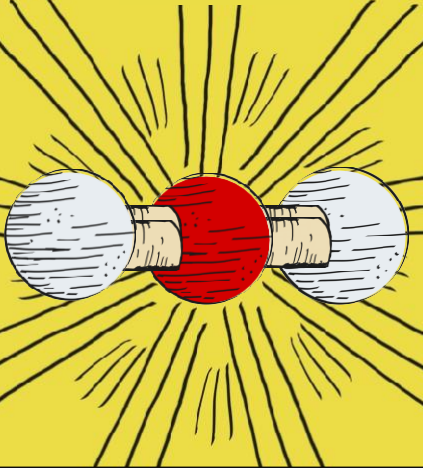
Carbon (IV) oxide

-395 ⚡ CO₂



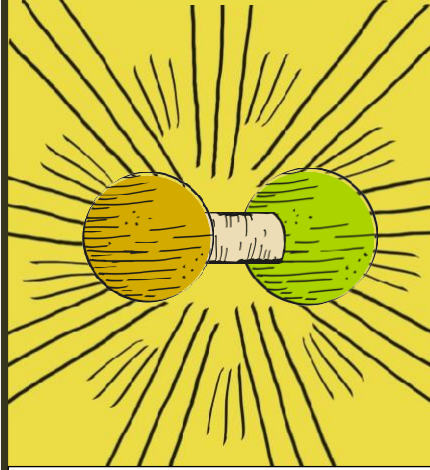
Carbon (IV) oxide

-395 ⚡ CO₂



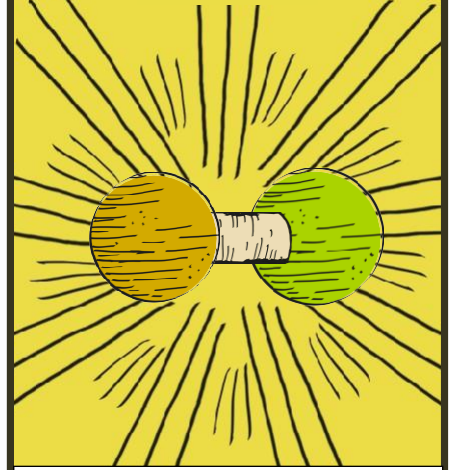
Carbon (IV) oxide

-95 ⚡ HCl



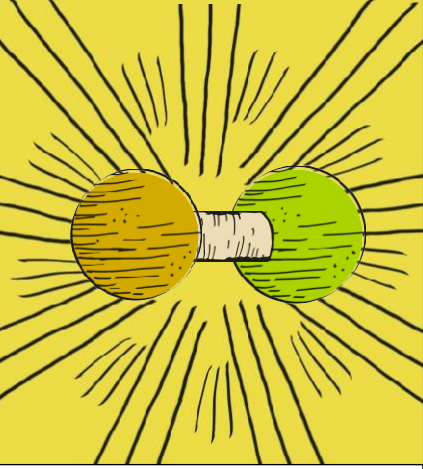
Hydrogen chloride

-95 ⚡ HCl



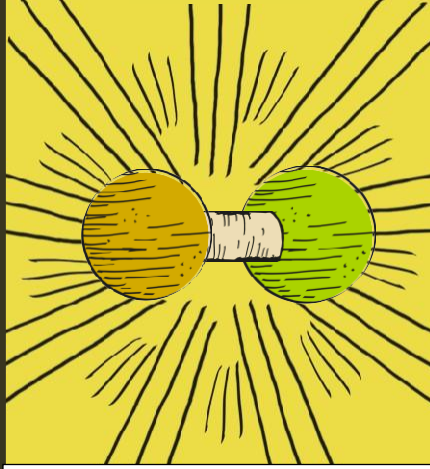
Hydrogen chloride

-95 ⚡ HCl



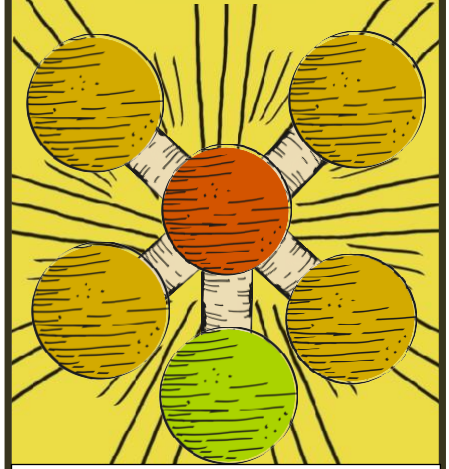
Hydrogen chloride

-95 ⚡ HCl



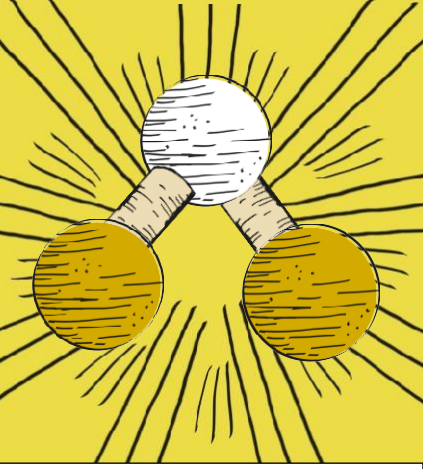
Hydrogen chloride

-315 ⚡ NH₄Cl



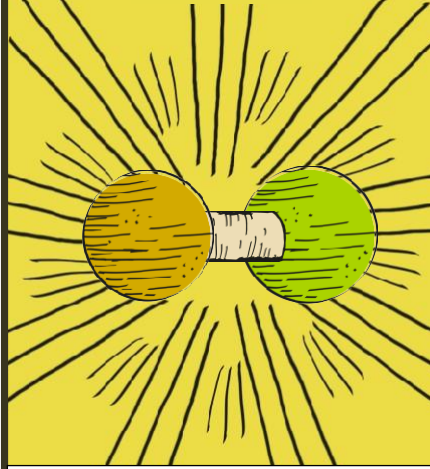
Ammonium chloride

-285 ⚡ H₂O



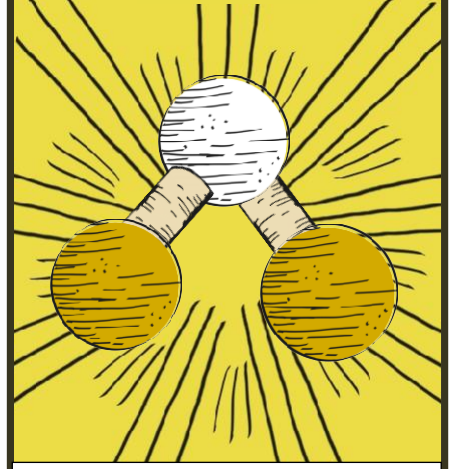
Water

-95 ⚡ HCl



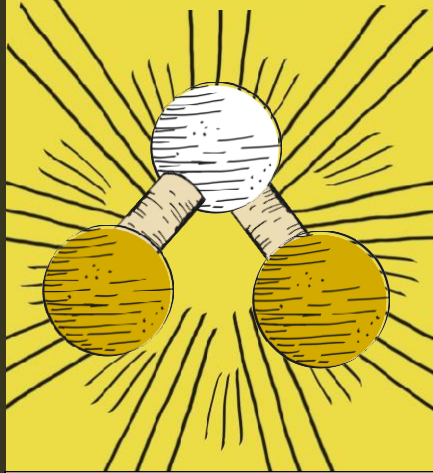
Hydrogen chloride

-285 ⚡ H₂O



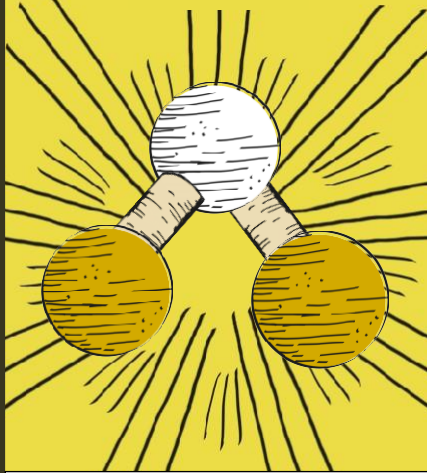
Water

-285 ⚡ H₂O



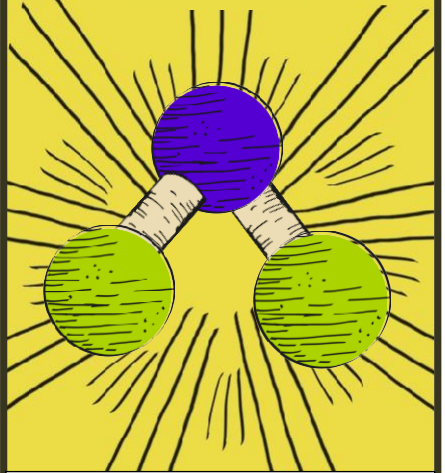
Water

-285 ⚡ H₂O



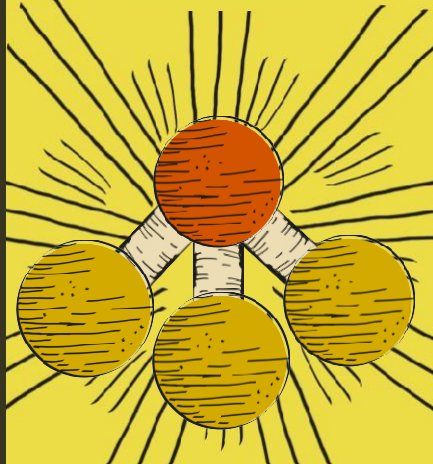
Water

-875 ⚡ CaCl₂



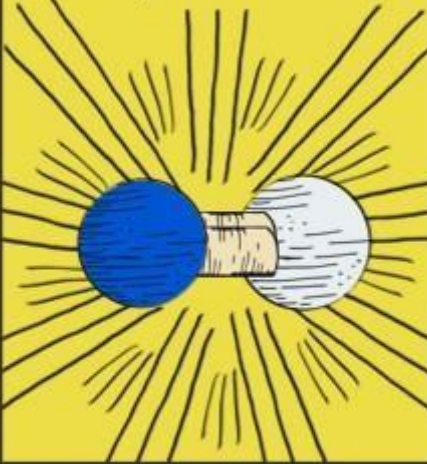
Calcium chloride

-45 ⚡ NH₃



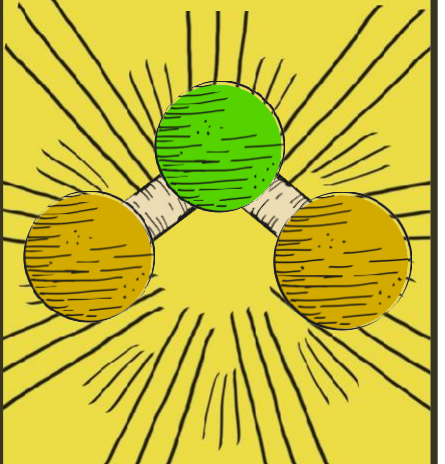
Ammonia

-120 ⚡ CaO



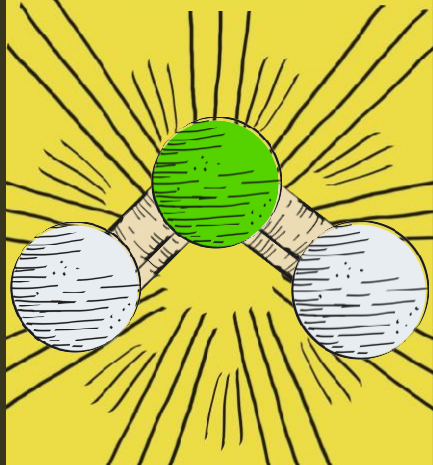
Calcium oxide

-20 ⚡ H₂S



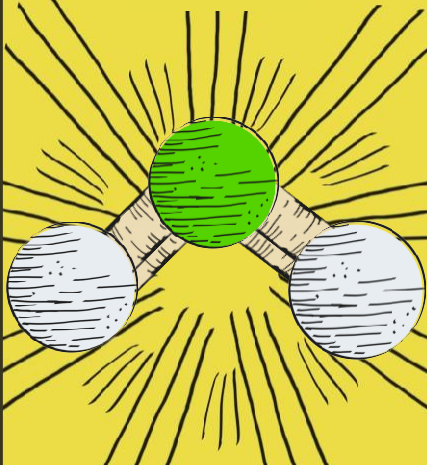
Hydrogen sulfide

-300 ⚡ SO₂



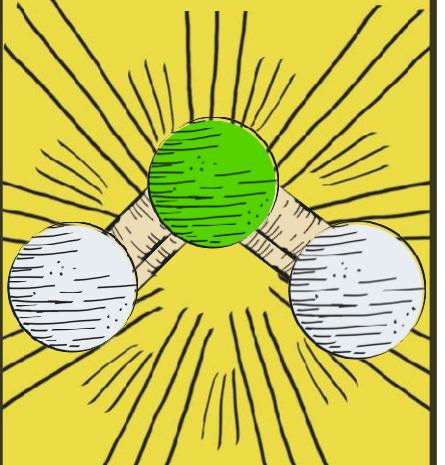
Sulfur (IV) oxide

-300 ⚡ SO₂



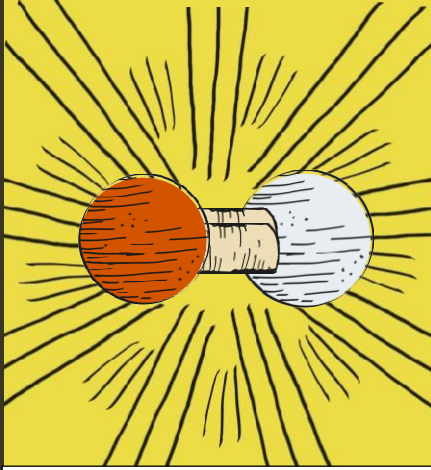
Sulfur (IV) oxide

-300 ⚡ SO₂



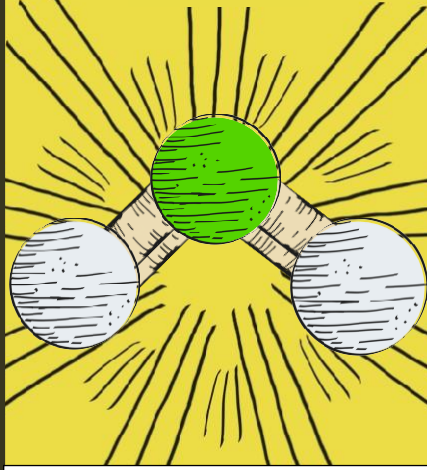
Sulfur (IV) oxide

+90 ⚡ NO



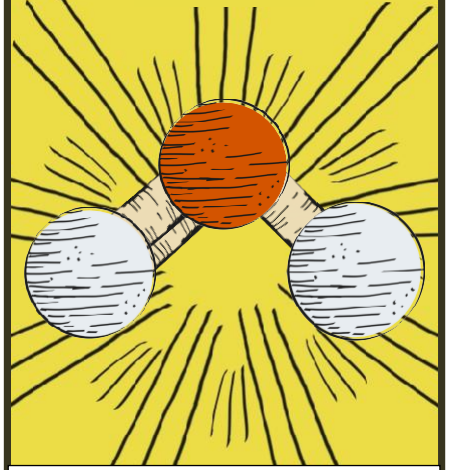
Nitrogen (II) oxide

-300 ⚡ SO₂



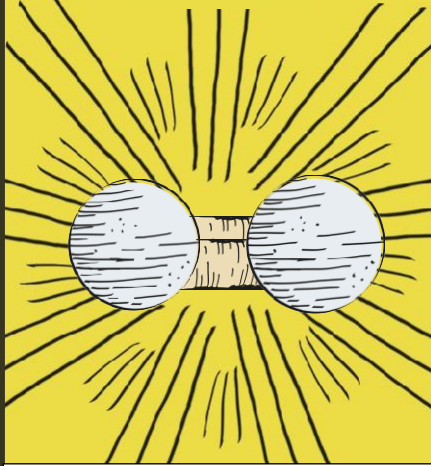
Sulfur (IV) oxide

+35 ⚡ NO₂



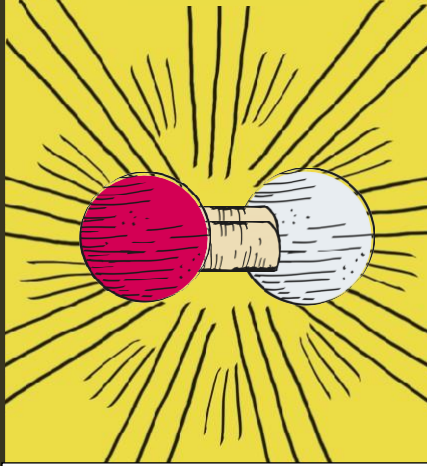
Nitrogen (IV) oxide

0 ⚡ O₂



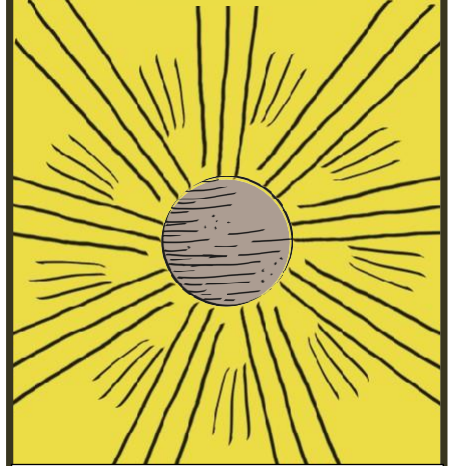
Oxygen

-600 ⚡ MgO



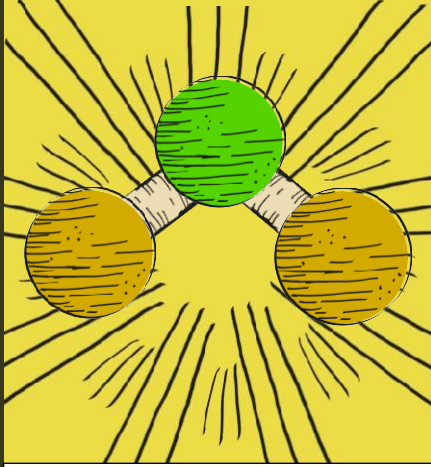
Magnesium oxide

0 ⚡ Fe



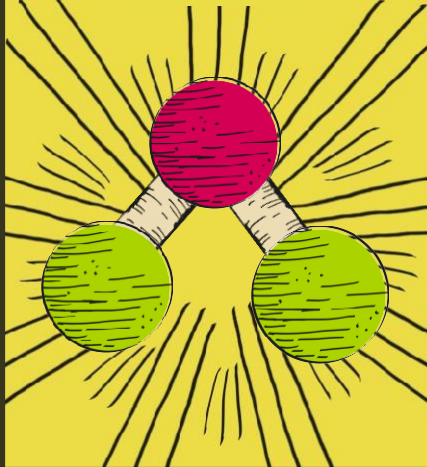
Iron

-20 ⚡ H₂S



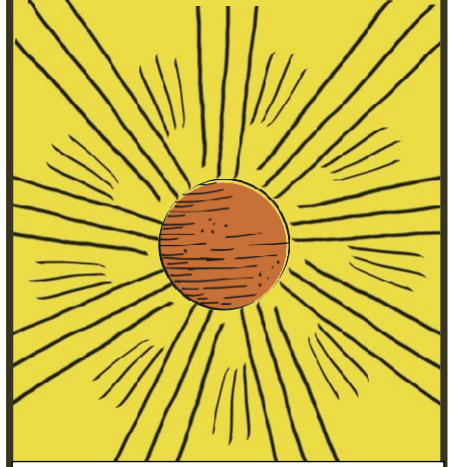
Vodonik - sulfid

-625 ⚡ MgCl₂

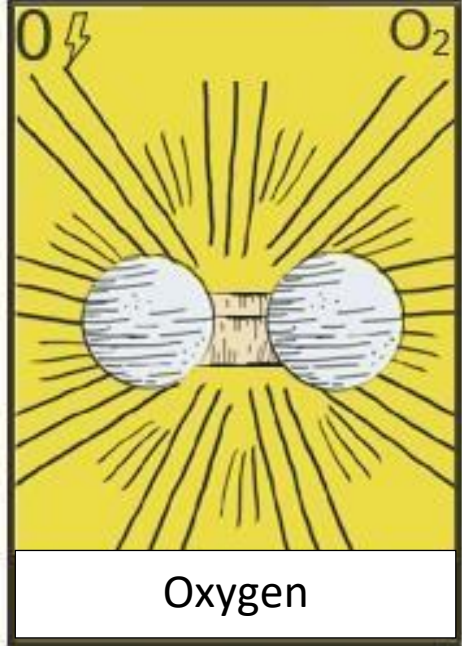
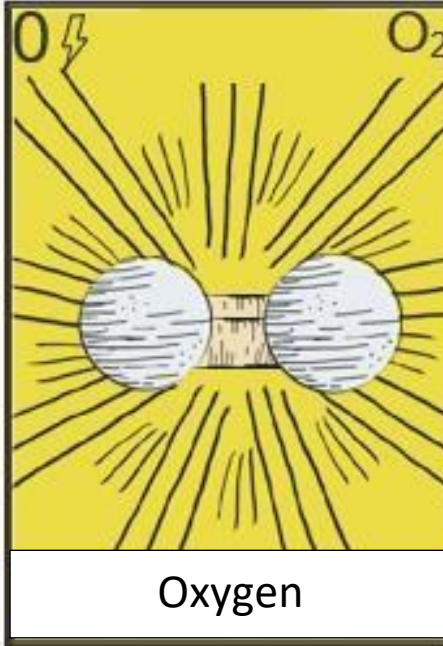
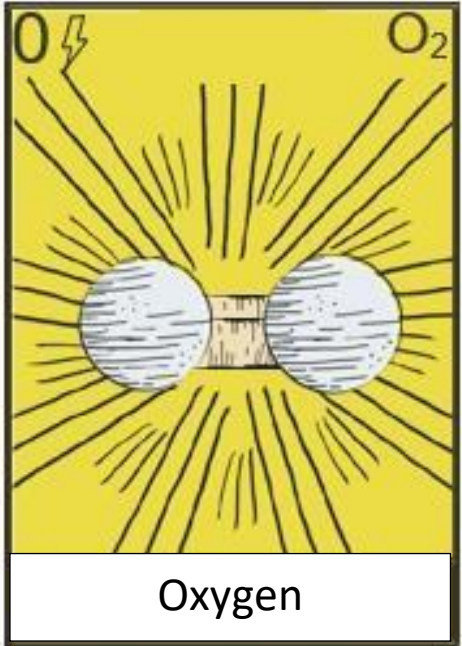
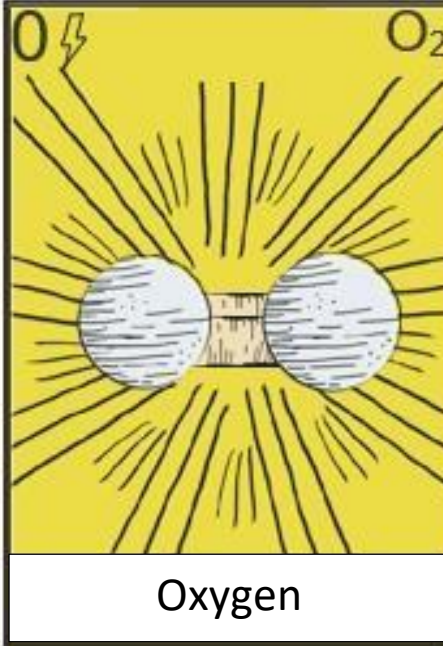
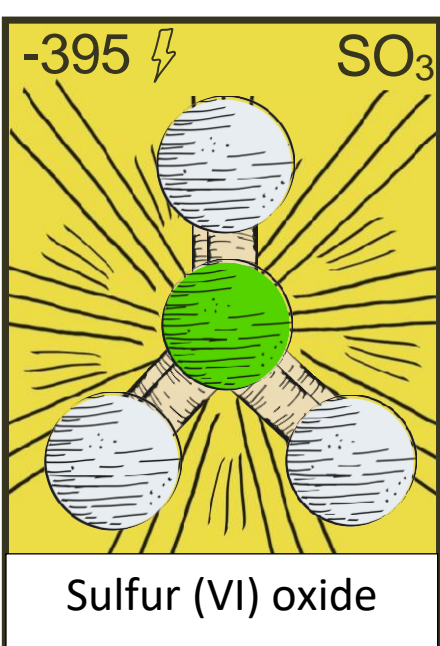
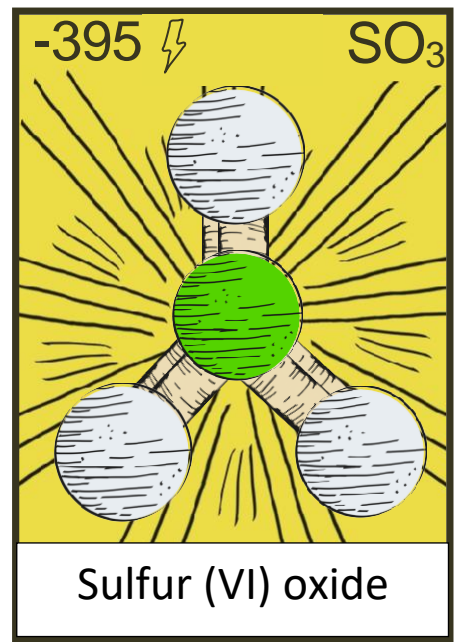
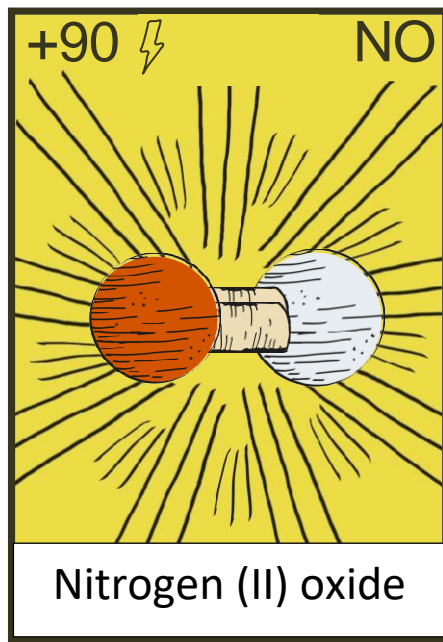
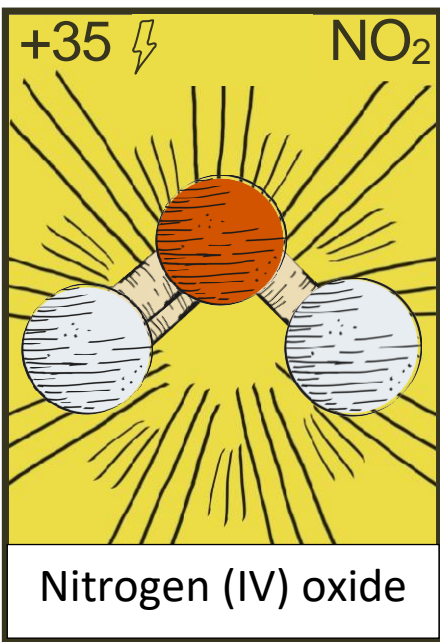


Magnesium chloride

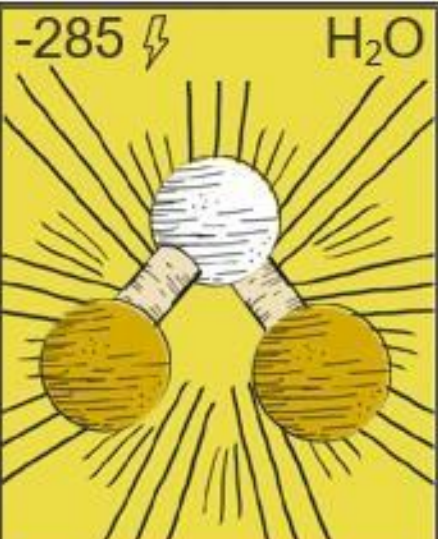
0 ⚡ Cu



Copper




-285 ⚡ H_2O



Water

The diagram shows a ball-and-stick model of a water molecule. It consists of one white sphere (oxygen) at the top, connected by two grey rods to two yellow spheres (hydrogen) at the bottom. The entire molecule is set against a yellow background with black radiating lines. A lightning bolt symbol is positioned to the left of the oxygen atom, and the chemical formula H₂O is in the top right corner.

0 ⚡ Fe



Iron

The diagram shows a single grey sphere representing an iron atom. It is set against a yellow background with black radiating lines. A lightning bolt symbol is positioned to the left of the sphere, and the chemical symbol Fe is in the top right corner.

